Chemistry Vocabulary: Measuring Matter



- 1. <u>matter</u>: the stuff that makes up everything in the universe. Anything that has mass and takes up space.
- 2. <u>properties (of matter)</u>: ways to describe or categorize matter. Hardness, texture, shape, temperature, flammability, size, and colore are all examples of properties of matter.
- <u>characteristic properties (of matter</u>): a quality of a substance that never changes and can be used to identify unknown substances (ex. Boiling point and melting point).
- 4. <u>chemical reactivity</u>: the ability of a substance to undergo a specific chemical change. This is a characteristic property.
- 5. <u>boiling point</u>: the temperature at which a substance boils (the boiling point of water is 212 degrees Fahrenheit or 100 degrees Celsius). This is a characteristic property.
- 6. <u>melting point</u>: the temperature at which a substance melts. This is a characteristic property.
- <u>density</u>: the measurement of how much mass of a substance is contained in a given volume. D = m/v
- 8. mass: a measure of how much matter is in an object.
- 9. weight: a measure of the force of gravity on an object.
- 10. volume: the amount of space that matter occupies.
- 11. <u>International System of Units ("SI"</u>): the system of units used by scientists to measure the properties of matter.
- 12. <u>states (of matter)</u>: the three traditional states of matter are solids (fixed shape and volume) and liquids (fixed volume and shaped by the container) and gases (filling the container); "the solid state of water is called ice."